# Studies on some Benzopyran Derivatives with Expected Antimicrobial and Antiviral Activity

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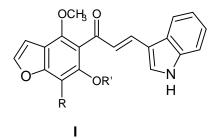
Abstract: The naturally occurring furocoumarin (xanthotoxin) (1) yielded 3-(6-hydroxy-7-methoxy-1-benzofuran-5-yl) acrylohydrazide (2) upon treatment with hydrazine hydrate. When 2 was treated with an equimolar amount of the appropriate isothiocyanates it gave the respective N-substituted – 2 - [3 - (6 - hydroxy - 7 - methoxy - 1 - benzofuran - 5 - yl) prop - 2 - enoyl] hydrazine carbothioamide (**3a-c**), which when heated with sodium hydroxide, yellow mercuric oxide and phosphorus oxychloride respectively were transformed into 5-[2-(4-substituted-5mercapto-4H-1,2,4-triazol-3-yl)vinyl]-7-methoxy-1-benzofuran-6-ol (**4a-c**), 5-{2-[5-(substituted amino)-1,3,4oxadiazol-2-yl] vinyl}-7-methoxy-1-benzofuran-6-ol (**5a-c**), and 5-{2-[5-(substituted amino)-1,3,4-thiadiazol-2-yl] vinyl}-7-methoxy-1-benzofuran-6-ol (**5a-c**) respectively. The reaction of **3a** with ethyl bromoacetate gave N'-(3benzyl-4-oxo-1,3-thiazolidin-2-ylidene)-3-(6-hydroxy-7-methoxy-1-benzofuran-5-yl)acrylo hydrazide (**7**). When **3a,b** was heated with ethyl cyanoacetate it yielded N-substituted-2-[4-(6-hydroxy-7-methoxy-1-benzofuran-5-yl)-2imino-2H-pyran-6-yl]hydrazine-carbothioamide (**8a,b**) respectively. The prepared compounds were tested for their antimicrobial and antiviral activities. [Nature and Science 2010;8(7):20-29]. (ISSN: 1545-0740).

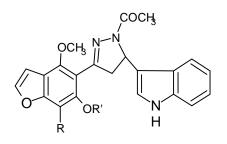
Key words: Benzopyran; Antimicrobial; Antiviral Activity

#### 1. Introduction

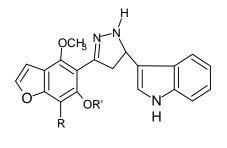
This work deals with the synthesis of some benzofuran derivatives derived from naturally occurring xanthotoxin for biological evaluation. A wide variety of pharmacological properties was shown to be associated with benzofuran derivatives<sup>(1)</sup> <sup>2)</sup>. Various thiosemicarbazides and their cyclized products e.g.triazoles, oxadiazoles, thiazolidinones and thiadiazoles are also associated with a broad spectrum of biological properties including  $anticonvulsant^{(3)}$ , anti-inflammatory<sup>(4-7)</sup>, antitumor, antiviral<sup>(8)</sup>. analgesic. ulcerogenic. lipid peroxidation<sup>(9)</sup>. antimicrobial<sup>(10)</sup> and anti-HIV

activities<sup>(11,12)</sup>. The chalcones **Ia-d**, pyrazolinyl derivatives **IIa-d**, N-acetyl pyrazolinyl derivatives **IVa-d** and isoxazolinyl derivatives **VIa-d** possess moderate activity towards the Gram +ve bacteria and yeast in a concentration of 200 $\mu$ g/disk compared with Chloramphenicol 10 $\mu$ g/disk. Also, these compounds possess slight activity towards Gram -ve bacteria used. On the other hand, when these compounds were subjected to the U.V, light (366 nm), the activity was found to be higher than those compounds not subjected to the U.V. light (<sup>13)</sup>.

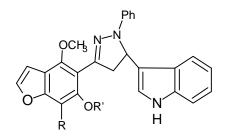


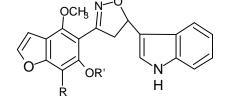












IV

VI

I,II,III,IV and VI a,R=R<sup>`</sup>=H b,R=OCH<sub>3</sub>;R<sup>`</sup>=H c,R=H;R<sup>`</sup>=CH<sub>3</sub>

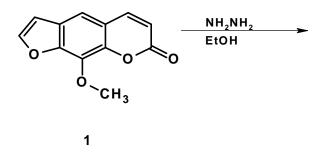
d,R=OCH<sub>3</sub>;R<sup>'</sup>=CH<sub>3</sub>

Figure 1. Furocoumarin Structure

H<sub>2</sub>N

# 2. Results and Discussion

Xanthotoxin (1) (9-Methoxy-7H-furo [3, 2-g] chromen-7-one) yielded 3-(6-hydroxy-7-methoxy-1-



benzofuran-5-yl) acrylohydrazide (2) upon treatment with hydrazine hydrate  $^{(14)}$ .

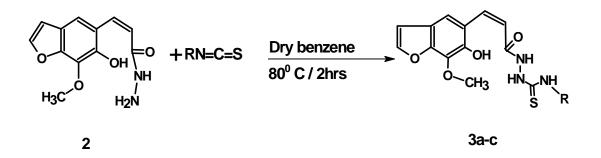
When compound **2** was treated with isothiocyanates derivatives namely, (benzyl-, ethyl- and cyclohexyl isothiocyanate), N-substituted–2-[3-(6–hydroxy–7–

methoxy-1- benzofuran-5-yl) prop-2-enoyl] hydrazido carbothioamides(**3a-3c**) were obtained.

Ò

СН₃

2



The IR spectra of compounds **3a-c** showed the presence of C=S group at 3063-3225 cm<sup>-1</sup>.

5-[2-(4-Substituted-5-mercapto-4H-1,2,4-triazol-3-

yl)vinyl]-7-methoxy-1-benzofuran-6-ols(**4a-4c**) were formed by the cyclization of compounds **3a-c** by heating in aqueous sodium hydroxide.

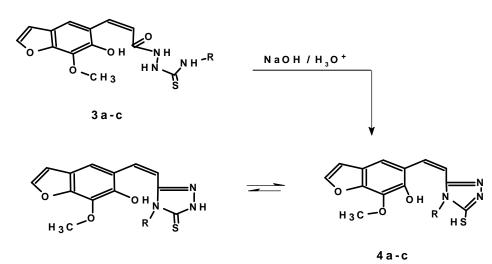
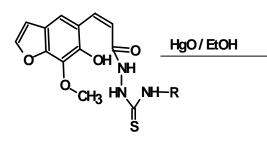


Figure 2.

The H<sup>1</sup>-NMR of compounds 4a-c confirmed the absence of the NH protons, and their IR spectra also showed the absence of the C=O and C=S groups. On the other hand desulfurization of compounds 3a-c



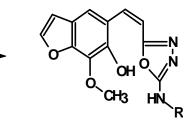
3a-c

The IR spectra of 5a-c revealed the absence of C=O and C=S absorption.

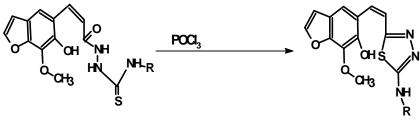
Treatment of compounds 3a-c with phosphorus

oxychloride and heating yielded the corresponding 5-

by yellow mercuric oxide in boiling ethanol yielded 5-{2-[5-(substituted amino)-1,3,4-oxadiazol-2yl]vinyl}-7-methoxy-1-benzofuran-6-ols (5a-c).



{2-[5 amino)-1,3,4-thiadiazol-2-yl]vinyl}-7methoxy-1-benzofuran-6-ols (6a-c).

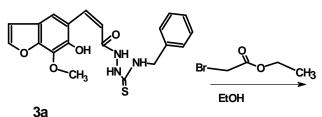


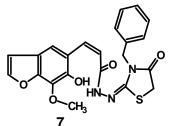
За-с

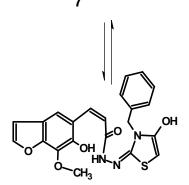
benzyl-4-oxo-1,3-thiazolidin-2-ylidene)-3-(6-hydroxy -7-methoxy-1-benzofuran-5-yl)acrylo hydrazide (7).

6а-с

The IR spectra of 6a-c revealed the absence of C=O and C=S absorption. The reaction of thiourea derivative 3a with ethylbromoacetate yielded N'-(3-



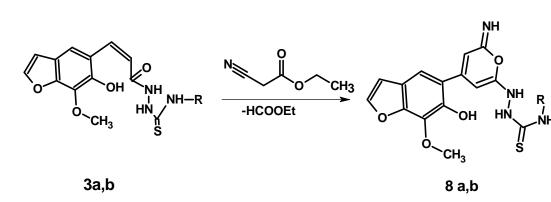




The IR spectra of 7 revealed the absence of C=S absorption.

The reaction of **3a,b** with ethylcyanoacetate yielded

N-substituted-2-[4-(6-hydroxy-7-methoxy-1benzofuran-5-yl)-2-imino-2H-pyran-6-yl] hydrazinecarbothioamide (**8a,b**).



a, R=PhCH<sub>2</sub>b, R=CH<sub>3</sub> CH<sub>2</sub>c, R=C<sub>6</sub>H<sub>11</sub>-

The IR spectra of **8a**, **b** revealed the absence of C=O absorption

### 3. Biology

### 3.1 Antimicrobial activity test

The antimicrobial activity screening test of compounds 3(a-c), 4a, 4c, 5b, 5c, 6a, 6c, 7, and 8a, b against G +ve, G -ve and fungi was carried out by using the disk-diffusion method with some modifications <sup>(15)</sup>. Whatman No.1 filter paper disks (0.5 cm) were sterilized by autoclaving at 121 °C for 15 min and then impregnated with the tested compounds (3 x 10<sup>-6</sup> g.or 300 µg/disk) (using chloroamphenical as control 300 µg/disk). The disks were placed on the surface of the cold solid medium in petri-dishes, inoculated with the tested microorganisms, then incubated at 5°C for permit good diffusion and then transferred to an incubator at 37 °C for 24 h for bacteria (1 and 2) and at 28°C for 24h for fungus (3) (cf. Table1).

# **3.2** Antimicrobial activity data conclusion of the data obtained in table (1)

The preliminary antimicrobial activity screening for the prepared compounds were tested using one local strain as G+ve, one local strain as G-ve bacteria and one local strain as fungus together with chloramphenicol as control.

From the data obtained (table 1) it is clear that only compounds 3c, 5c (thiourea and oxadiazole derivatives) were found to be highly active against G+ve bacteria compared with the reference antibiotic chloramphenicol.

Compounds 3a and 8b also possess moderate activity towards G+ve bacteria while the other compounds are slightly active towards the same organism. On the other hand, compounds 3c and 5c possess moderate activity towards G-ve bacteria compared with the control, while the other compounds possess slight activity towards the same organism. Moreover, compounds 3a, 3b, 4c, 6c, 8a and 8b showed moderate activity towards the fungus compared with the reference, while the other compounds possess slight activity towards the same organism.

In general, only compounds 3c and 5c showed high activity towards G+ve bacteria and moderate activity towards G-ve bacteria.

**N.B** the activity of our target compounds, namely triazole, oxadiazole, thiadiazole, thiazolidine and pyrane derivatives will be studied in details against clinical pathogenic microorganisms and also their MIC will be published separately in the near future.

# 3.3 Antivirus activity

# Cells and viruses

Herpes Simplex Virus type 1(HSV-1) was used for antiviral bioassay. The virus was isolated and propagated in the Virology Laboratory, Department of Water Pollution Researches, and National Research Center.

African green monkey kidney cells (Vero) was used for HSV-1 propagation and bioassay. Cells were grown in minimum essential medium and supplemented by 1% antibiotic-antimycotic mixture (GIBCO-BRL), 8% fetal bovine serum (Sigma) and the pH was adjusted to (7.2-7.4) by 7.5% sodium bicarbonate solution. Cells were grown as monolayer sheets and dissociation by trypsin-versene solution (0.15% trypsin and 0.04% ethylene diamine tetraacetic acid, EDTA 2Na). The dissociated cells were subcultured in a 96-well plate to measure the cvtotoxicity of the prepared compounds (Silva et al., 1997). Cytotoxicity assay were carried out for the prepared samples to determine the safe concentrations to be used for antiviral bioassay.

# Plaque infectivity reduction assay (PIRA):

A 6-well plate was cultivated with Vero cell culture (105cell/ml) and incubated for 2 days at 370 C. The virus was diluted to give 107 PFU/ml final concentrations and mixed with the tested compounds at different concentrations and incubated overnight at 40• C. Growth medium was removed from the multiwell plates and virus-compound mixtures were inoculated (100µl /well). After 1h contact time, the inoculum was aspirated with 3ml of MEM and 1% agarose overlaid the cell sheets. The plates were left to solidify and then incubated at 37•C until the development of virus plaques. Cells were fixed in 10% formalin solution for 2hr, and stained with crystal violet stain. Control virus and cells were treated identically without compounds. Virus plaques were counted and the percentage of reduction was calculated (Silva et al., 1997).

# 4. Experimental

All melting points were uncorrected. Elemental analysis was carried out in the microanalytical unit of the National Research Centre. **IR** spectra were recorded on a Mattson-5000 FTIR spectrometer using KBr wafer technique.<sup>1</sup>**H-NMR** spectra were determined on a varian-Gemini-300 MHz. And Jeol-Ex-300 MHz NMR spectrometer using TMS as an internal standard with (chemical shift. = 0 ppm). Mass spectra were determined on Finnigan mat SSQ 7000 mode: EI,70Ev (Thermo Inst.Sys.Inc.,USA). The purity of the synthesized compounds was tested by thin layer chromatography (TLC), Merck plates.

4.1. Synthesis of N-substituted – 2 - [-3 -(6– hydroxy – 7 – methoxy – 1 –benzofuran – 5 - yl) prop– 2 -enoyl] hydrazine carbothioamides (3a-c). To a suspension of 2 (0.01 mol) in dry benzene (50 ml), the appropriate isothiocyanate (0.01 mol) was added. The reaction mixture was heated at 80<sup>o</sup>C with stirring for 2 hrs and then left overnight at room temperature. The solid so obtained was filtered off and crystallized from ethanol to give **3a-c** (cf.Table 2).

# 4.2. Synthesis of 5-[2-(4-substituted-5-mercapto-4H-1, 2,4-triazol-3-yl) vinyl]-7-methoxy-1benzofuran-6-ols (4a-c).

Compound of **3a-c** (0.01mol) in sodium hydroxide (5ml, 2N) was refluxed under stirring for 10 hours. The reaction mixture was then cooled and neutralized with dilute hydrochloric acid. The precipitate thus obtained was filtered off, washed with water several times, dried and crystallized from ethanol to give **4a-c**(cf.Table 2).

## **4.3.** Synthesis of 5-{2-[5-(substituted amino)-1,3,4oxadiazol-2-yl] vinyl}-7-methoxy-1-benzofuran-6ols (5a-c).

Compounds of 3a-c (0.002 mol) were refluxed with excess yellow mercuric oxide (0.015 mol) in ethanol

(30ml) for 4-6 hours. The reaction mixture was allowed to cool to room temperature (to allow the sedimentation of the black mercuric sulphide),filtered and mercuric sulphide was washed with ethanol. The filtrate and alcoholic washing were combined, treated with water until a permanent turbidity existed and allowed to stand overnight .The product was separated and crystallized from ethanol to give **5a-c** (cf. Table 2).

# 4.4. Synthesis of 5-{2-[5-(substituted amino)-1,3,4-thiadiazol-2-yl] vinyl}-7-methoxy-1-benzofuran-6-ols (6a-c).

Phosphorous oxychloride (15ml) was added to the appropriate compound **3a-c** (0.005 mol) and the mixture was heated under reflux for 2-4 hours. The mixture was then evaporated in vacuo and the residue was washed with dilute ammonium hydroxide solution and water, dried and crystallized from ethanol to give **6a-c**(cf.Table 3).

# 4.5. Synthesis of N'- (3-benzyl-4-oxo-1, 3-thiazolidin-2-ylidene)-3-(6-hydroxy-7-methoxy-1-benzofuran-5-yl)acrylo hydrazide (7).

A mixture of **3a** (0.01 mol), ethyl bromoacetate

(0.01 mol) and anhydrous sodium acetate (0.015 mol) in absolute ethanol (30ml) was refluxed for 3h. The reaction mixture was cooled, diluted with water and allowed to stand overnight. The solid so obtained was filtered off, dried and crystallized from ethanol to give 7 (cf.Table 3).

# 4.6. Synthesis of N-substituted-2-[4-(6-hydroxy-7-methoxy-1-benzofuran-5-yl)-2-imino-2H-pyran-6-yl]hydrazinecarbothioamides (8a,b).

A mixture of 3a,b (0.01 mol) and ethyl cyanoacetate (0.01 mol) in ethanol (30 ml) containing few drops of glacial acetic acid was refluxed for 4 hours. The precipitate that formed was filtered off, dried and crystallized from the proper solvent to give 8a,b(cf.Table 3).

## 5. Results and Discussion

Cytotoxicity was carried out to determine the non cytotoxic doses of the prepared compounds to be used for antivirus bioassay. The results showed that no toxic effects were observed for all samples at dilution > 1:8 which permit the safe use of different concentrations of the tested materials for antiviral bioassays.

The results revealed that oxadiazole carboxylic acid was of highest activity against HSV-1 than all the other tested materials. It was noticed that cyclohexyl group increased the potential of some tested compounds for anti-HSV-1 activity as shown in the case of cyclohexyl thiadiazole, cyclohexyl triazole, cyclohexyl oxadiazole as shown in fig (A,B and C, respectively).

Sample No.								
	Micro-organism							
	G + ve (1)	G – ve (2)	Fungi (3)					
Control	++++	++++	++++					
3a	++	+	++ ++					
3b	+	+						
3c	+++	++	+					
4a	+	+	+					
4c	+	+	++					
5b	+	+	+					
5c	+++	++	+ +					
6a	+	+						
6c	+	+	++					
7	+	+	+					
8a	+	+	++ ++					
8b	++	+						
Micro-organisms:								

#### Table 1. The prelininary antimicrobial screening test for the prepared compounds using chloroamphenical as control

Micro-organishis:

1-Bacillus Subtilies (G + ve)

2- Escherichia Coli (G - ve)

3-Aspergillus Niger (Fungus)

The inhibition zones were measured in the following manner:

Inhibition zone +++= highly active (> 12 mm) Inhibition zone ++= moderately active (9-12 mm) Inhibition zone += slightly active (6-9 mm)

Inhibition zone - = non sensitive (0.5 mm)

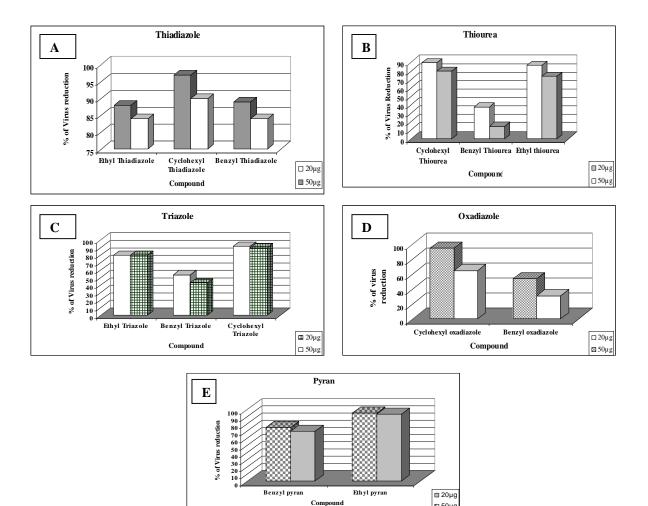
Compound no.	Mp ( <sup>0</sup> C) Yield(%)	Molecular formula (M.wt)	% Analysis Calcd/(Found)			IR( ,cm <sup>-1</sup> ) and <sup>1</sup> H-NMR (DMSO-d <sub>6</sub> )	MS (M <sup>+</sup> )
		× ,	С	H	N	-	(1.1)
<b>3</b> a	198-200 (95)	C <sub>20</sub> H <sub>19</sub> N <sub>3</sub> O <sub>4</sub> S (397.449)	10.57 10.42	4.82 4.72	60.44 60.34	IR: 1335 (C=S), 3063-3255 (3NH), 3343 (OH). <sup>1</sup> H-NMR: 3.99 (s, 3H,OCH <sub>3</sub> ), 4.72 (d, 2H,CH <sub>2</sub> Ph), 7.25 (d, 1H furan H-3, J <sub>H,H</sub> = 2.3 H <sub>Z</sub> ), 7.87 (d, 1H furan H-2, J <sub>H,H</sub> = 2.3 H <sub>Z</sub> ), 7.31-7.36 (m, 8H,aromatic protons+2H	397
3b	>300 (85)	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>4</sub> S (335.379)	12.53 12.46	5.11 5.05	53.72 53.65	olefinic protons), 7.41- 7.82 (3br.s, 3H,3NH). IR: 1305(C=S),3128-3204 (3NH), 3250(OH). <sup>1</sup> H-NMR: 1.08 (t,3H,CH <sub>3</sub> ),3.50 (m,2H,CH <sub>2</sub> ), 7.79-7.82 (3br.s,3H,3 NH), 9.06 (br.s,1H,OH).	335
3c	217-220 (85)	C <sub>19</sub> H <sub>23</sub> N <sub>3</sub> O <sub>4</sub> S (389.470)	10.79 10.68	5.95 5.88	58.59 58.43	$\begin{array}{l} \mbox{IR: } 1341(C=S), 3138-3013 \ (3NH) \ and \ 3243 \\ \ cm^{-1} \ (OH). \ ^1H-NMR: \ 3.98 \ (s, 3H, OCH_3), \\ 1.27-2.21 \ (m, 10H \ of \ cyclohexyl), \ 4.07 \ (m, 1H \ of \ cyclohexyl), \ 6.82 \ and \ 7.75 \ (2d, 2H \ furan \ H-3, H-2, J_{H,H}=2.3Hz), 7.80-7.82 \ (3br.s, 3H, 3NH) \\ , \ 6.83 \ and \ 6.99 \ (2d, 2H \ olefine \ protons \ ), 9.13 \ (br.s, 1H, OH), \end{array}$	389
<b>4</b> a	>300 (95)	C <sub>20</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub> S (379.433)	11.07 11.00	4.52 4.46	63.31 63.20	IR: 1619(C=N),3238(OH). <sup>1</sup> H-NMR: 3.98 (s, 3H,OCH <sub>3</sub> ), 7.25 (d, 1H furan H-3, $J_{H,H}$ = 2.3 Hz), 7.88 (d, 1H furan H-2, $J_{H,H}$ = 2.3 Hz), 7.29-7.33 (m, 5H,aromatic protons and 2H	379

Table 2. Physical and spectral data of the newly synthesized compounds (3a-c)-(5a,b).

4b	205-210 (98)	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> O <sub>3</sub> S (317.364)	13.24 13.11	4.76 4.65	56.77 56.67	olefinic protons), 12.75 (s, 1H,SH). IR: 1630(C=N),3238(OH) . <sup>1</sup> H-NMR: 1.02(t,3H,CH <sub>3</sub> ),2.50(q,2H,CH <sub>2</sub> ),3.92(s,3H,O CH <sub>3</sub> ),6.8-7.81 (m,5H,aromatic protons + olefinic protons ), 12.6 (br.s, 1H, OH) , 13.4(s,1H,SH).	317
4c	170-174 (98)	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> S (371.454)	11.31 11.21	5.70 5.65	61.44 61.32	IR: $1636(C=N),3347(OH)$ . <sup>1</sup> H-NMR: 3.3(m,1H,cyclohexyl),1.03-1.77 (m,10H,cyclohexyl), the OH signal gave one D <sub>2</sub> O exchangeable signal at 9.14.	371
5a	196-201 (60)	C <sub>20</sub> H <sub>17</sub> N <sub>3</sub> O <sub>4</sub> (363.367)	11.56 11.46	4.72 4.61	66.11 66.03	IR: 1670(C=N), 3323(NH), 3343(OH). <sup>1</sup> H-NMR:	363
5b	214-216 (65)	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> O <sub>4</sub> (301.297)	13.95 13.82	5.02 4.96	59.79 59.68	IR: 1623(C=N),3349 (NH), 3396(OH).	301
5c	208-212 (60)	$\begin{array}{c} C_{19}H_{21}N_3O_4\\ (355.388) \end{array}$	11.82 11.74	5.96 5.87	64.21 64.10	IR: 1656(C=N),3267 (NH), 3304(OH).	355

Table 3. Physical and spectral data of the newly synthesized compounds (6a-c),7, and (8a,b).

Compound no.	Mp ( <sup>0</sup> C) Yield(%)	Molecular formula (M.wt)	% Analysis Calcd/(Found)		Found)	IR( ,cm <sup>-1</sup> ) and <sup>1</sup> H-NMR (DMSO-d <sub>6</sub> )	MS (M <sup>+</sup> )
		(111. 111)	С	Н	Ν		
6a	185-190 (78)	$\begin{array}{c} C_{20}H_{17}N_3O_3S\\ (379.433) \end{array}$	11.07 10.95	4.52 4.43	63.31 63.26	$\begin{array}{llllllllllllllllllllllllllllllllllll$	379
6b	>300 (60)	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> O <sub>3</sub> S (317.364)	13.24 13.16	4.76 4.67	56.77 56.67	(br.s,1H,OH). IR: 1622(C=N),3140 (NH) , 3387(OH). <sup>1</sup> H-NMR: 1.11 (t,3H,CH <sub>3</sub> ), 2.49 (m,2H,CH <sub>2</sub> ) , 4.00 (s,3H,OCH <sub>3</sub> ), 6.84- 7.84 (m,3H,aromatic protons +2H olefinic protons).	317
6с	185-191 (65)	C <sub>19</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> S (371.454)	11.31 11.21	5.70 5.65	61.44 61.39	IR: 1634(C=N),3188 (NH) , 3241(OH). <sup>1</sup> H-NMR: 1.22-1.9(m,10H,cyclohexyl),3.43 (m,1H,cyclohexyl),3.99 (s,3H,OCH <sub>3</sub> ),6.5-7.9 (m,5H,aromatic protons + olefinic protons).	371
7	224-229 (75)	$\begin{array}{c} C_{22}H_{19}N_3O_5S\\ (437.470) \end{array}$	9.61 9.52	4.38 4.21	60.40 60.29	IR: 1609 (C=N),1712 (C=O), 3031(NH). <sup>1</sup> H-NMR: 4.79 (s,2H,CH <sub>2</sub> Ph),7.25-7.38 (m,8H,aromatic protons+2H olefinic protons).	437
8a	192-199 (75)	$\begin{array}{c} C_{22}H_{20}N_4O_4S\\ (436.485) \end{array}$	12.84 12.92	4.62 4.75	60.54 60.61	IR:1353 (NC=S), 3028-3253 (NH). $^{1}$ H-NMR:4.7 (d,2H,CH <sub>2</sub> Ph) ,5.7-8.6 and 9.8 (br.s,4H,4NH),9.5(br.s,1H,OH),7.0-7.6 (m,8H,aromatic protons + 1H olefinic proton ).	436
8b	209-213 (75)	$\begin{array}{c} C_{17}H_{18}N_4O_4S\\ (374.415) \end{array}$	14.96 14.82	4.85 4.75	54.53 54.61	IR: 1368 (NC=S), 3019-3280 (NH). <sup>1</sup> H- NMR:1.21 (t,3H,CH <sub>3</sub> ) ,3.46 (q,2H,CH <sub>2</sub> ), 6.8-7.87 (m,3H aromatic protons + 1H olefinic proton),9.06-9.11 (4br.s,4H,4NH).	374



Anti-HSV 1 activity of the selected prepared compounds. A. Thiadiazole (6a-c) B. Thiourea (3a-c) C. Triazole (4a-c) D.Oxadiazole (5a,c) E.Pyran (8a,b)

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